

# Revisiting the original reasons for excluding inorganic fertilizers in organic farming—Why the ban is not consistent with our current scientific understanding

Holger Kirchmann 

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## Abstract

This paper reviews the original reasons of the organic farming movement for excluding mineral (inorganic) fertilizers. In this paper, their theories and decision criteria for excluding use of inorganic fertilizers in crop production were revisited. Original reasons for banning inorganic fertilizers were subjected to scientific scrutiny, which was not possible when they were formulated 50–100 years ago due to limited knowledge of the soil-crop system. The original reasons were as follows: Rudolf Steiner, the founder of biodynamic farming, played down the physical role of plant nutrients and pointed out “flow of forces” as being most important for soils and crops. Eve Balfour and Albert Howard, founders of the Soil Association in England, claimed that inorganic fertilizer increases the breakdown of humus in soil, leading to a decline in soil fertility. Hans-Peter Rusch, the founder of biological organic farming, considered inorganic fertilizers to be imbalanced products not matching crop composition and not in synchrony with crop demand. When testing these historical statements as scientific hypotheses, older and modern scientific literature was used for validation. Steiner’s belief about the “flow of forces” has not been verified using current methodologies. The claim by Balfour and Howard that inorganic fertilizers accelerate soil organic matter decomposition is not substantiated by data from long-term field experiments on carbon and nitrogen cycling in soil-plant systems. The statement by Rusch that inorganic fertilizers supply crops inappropriately is difficult to uphold, as the composition, time, and rate of application and the placement of fertilizer in soil or on foliage can be fully adapted to crop requirements. In light of accumulated scientific evidence, the original arguments lack validity. The decision to ban inorganic fertilizers in organic farming is inconsistent with our current scientific understanding. Scientific stringency requires principles found to be erroneous to be abandoned.

## Keywords

Steiner, Balfour, Rusch, view on nature, test of core statements

## Introduction

Exclusion of inorganic fertilizers is often perceived as a quality-improving principle that makes organic farming superior to conventional agriculture. The ban on the use of inorganic fertilizers in organic farming was introduced almost a century ago, when scientific knowledge about interactions between fertilizers and the soil-crop system was limited. This paper analyzes the original arguments in the different schools of organic agriculture on prohibiting the use of inorganic fertilizers and re-evaluates these reasons based on accumulated science and evidence. As inorganic fertilizers are the most powerful management input affecting the performance and quality of crops, it is highly important to gain a correct scientific understanding and go beyond popular biases against the use of inorganic fertilizers.

The analysis consisted of the following steps: First, original statements by the initiators of biodynamic farming (Steiner, 1924), the Soil Association (Balfour, 1943; Howard,

1940), and biological organic farming (Rusch, 1978) were identified and cited. The reasons cited for not using mineral fertilizers were then treated as scientific hypotheses, following the principles of the philosophy of science (Popper, 1959), and tested for their validity. The principal question addressed was whether there is evidence supporting the decision to ban the use of synthetic fertilizers in agriculture, i.e. whether the doctrine is corroborated by science.

At the time when exclusion of inorganic fertilizers was proposed, it was difficult to corroborate or reject the specific reasons put forward. There was a lack of knowledge

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Department of Soil and Environment, Swedish University of Agricultural Sciences, Uppsala, Sweden

### Corresponding author:

Holger Kirchmann, Department of Soil and Environment, Swedish University of Agricultural Sciences, Box 7014, 75007 Uppsala, Sweden.  
Email: holger.kirchmann@slu.se

**Table 1.** Original reasons postulated by the initiators of the organic farming movement for banning inorganic fertilizers based on their view on nature.

Initiator	Reasoning about inorganic fertilizers	Theory	View on nature
R. Steiner, 1861–1925	Inorganic fertilizers disrupt the flow of “cosmic and terrestrial forces” in soil-plant systems and greatly decrease crop quality	“Forces” captured and transferred into crops are the most important quality variables	There is a spiritual world hidden in nature. <i>Anthroposophy</i>
A. Howard, 1873–1947 E. Balfour, 1899–1990	Inorganic fertilizers speed up humus breakdown in soil	Only humus guarantees healthy crops and thereby healthy animals and humans	Nature is pure, ideal and kind. <i>Romanticism</i>
H.-P. Rusch, 1906–1977	Inorganic fertilizer cannot mimic the natural supply of nutrients by soils, so crops are fed inappropriately	Principles observed in nature must guide agriculture. Nature is a perfect unit with equal value of all living things	Human activities should be restricted to balance relations with nature <i>Environmentalism</i>

about soils and about nutrient turnover in the soil-plant system. Furthermore, scientific methodologies were limited or had not been developed, e.g. isotope techniques were not available, measurements of nutrient leaching and gaseous emissions were scarce, and an understanding of soil biological processes was only in its infancy. Thus it was difficult to prove the validity of the original statements when they were first made.

### Founders of organic farming were concerned about maltreatment of nature and deteriorating food quality

Two initiators of organic farming, the Austrian philosopher Dr. R. Steiner (1861–1925), who founded biodynamic farming, and the British agronomist and farmer Lady E. Balfour (1899–1990), who co-founded the Soil Association, were convinced that food quality had declined through the use of inorganic fertilizers. Their explicit response was to exclude inorganic fertilizers, in order to produce food of high quality. Steiner believed that food products would degenerate to such an extent that they would not be suitable as food for humans within a century (Steiner, 1924: 12). Balfour believed that if the fertility of soils is built up with an adequate supply of humus, crops do not suffer from diseases and animals fed on these crops develop high resistance to diseases. She also believed that humans fed with such plants and animals could achieve a high standard of health and the power to resist diseases and infections, from whatever cause (Balfour, 1943: 91). She was convinced that there must be something lacking in the foods of the day that was not lacking in the foods of the nation’s more robust forefathers (Balfour, 1943: 35). Her view was shared by the British agronomist Sir Albert Howard (1873–1947), her co-founder in the Soil Association, who stated that perfectly healthy soils are the basis for health on earth and that undernourishment of the soil is at the root of all problems (Howard, 1947: 12).

The German medical doctor H.-P. Rusch (1906–1977), the initiator of biological organic farming, believed that nature must be treated as a biological wholeness (Rusch, 1978: 15). He taught that each unnatural intervention in biological element cycling will cause damage (Rusch,

1978: 235); that each nitrogen fertilization prevents biological cycling and decreases biological quality (Rusch, 1978: 236); and that the quality of food is dependent on the biological functioning of soils (Rusch, 1978: 26), through proper humus management (Rusch, 1978: 234).

The different views on nature held by the initiators of the organic farming movement are summarized in Table 1. Note that the initiators did not argue against the use of inorganic fertilizers due to concerns about resource shortage, energy demand for nitrogen fertilizer production, or environmental impacts. It is only more recently that followers of the organic movement have declared that inorganic fertilizers are the root cause of eutrophication (e.g., Granstedt, 2000; Granstedt et al., 2008; Koepf, 1973) and that exclusion of inorganic fertilizers decreases greenhouse gas emissions (e.g., Muller et al., 2017; Smith et al., 2019).

### Testing Steiner’s statements about inorganic fertilizers

Steiner’s context of thinking was non-scientific. He described “auras” and “forces” around organisms, phenomena that are not known to science. His perspective was spiritual teaching, mysticism, and esoteric wisdom. He applied his spiritual perception to transform society, e.g. arts and architecture, medicine, religion, pedagogics, and also agriculture and bee-keeping. He created a wide-ranging spiritual system called “anthroposophy.”

Biodynamic agriculture builds upon a series of lectures by Steiner during a 1-week course on agriculture in Koberwitz (now Wrocław), Poland, in 1924. There, he taught a group of followers on consideration of spiritual matters in agriculture, in a lecture entitled “Geisteswissenschaftliche Grundlagen zum Gedeihen der Landwirtschaft” (Spiritual foundations for the renewal of agriculture), with instructions on how to produce organic food supplying “forces” to mankind (Steiner, 1924). This was the birth of the first distinct form of organic agriculture. Key statements from his book are as follows:

“Es weiss zum Beispiel kein Mensch heute, dass alle die mineralischen Dungarten gerade diejenigen sind, die zu dieser Degenerierung, von der ich gesprochen habe, zu diesem

Schlechterwerden der landwirtschaftlichen Produkte das Wesentliche beitragen" (*Nobody knows today that all sorts of mineral fertilizers are leading to the degeneration, of which I have talked, to the significant quality decline in agricultural products*). (Steiner, 1924: 20)

"Man hört heute sehr oft die Phrase: der Dünger enthalte die Futterstoffe für die Pflanzen. Da muss sie (Wissenschaft) sich korrigieren, weil eben sie eben von einer ganz falschen Anschauung ausgeht in Bezug auf die Ernährung irgendeines Wesens" (*You often hear the phrase: Fertilizer contains the nutrients for plants. In this case, science must correct itself, because this is a completely wrong view about nutrition of any organism*). (Steiner, 1924: 87)

"Man muss wissen, dass das Düngen zu einer Verlebendigung der Erde bestehen muss . . ." (*You need to know that fertilization means making the soil more alive*). (Steiner, 1924: 91)

"Für die Pflanze sind viel wichtiger lebendige Kräfte als bloss die substanzialen Kräfte, als bloss die Substanzen" (*For the plant, living forces are much more important than only substances*). (Steiner, 1924: 124)

"Daher werden Pflanzen, welche unter dem Einfluss irgendwelchen mineralischen Düngern stehen, ein solches Wachstum zeigen, das verrät, wie wie es nur unterstützt wird von angeregter Wässrigkeit, nicht von lebendiger Erdigkeit" (*Therefore plants treated with mineral fertilizers will show growth affected by waterishness, not living soil*). (Steiner, 1924: 94)

"Man muss die Erde direkt beleben, und das kann man nicht, wenn man mineralisierend vorgeht . . ." (*One needs to revive the soil, and this cannot be done using minerals . . .*). (Steiner, 1924: 122)

"Denn jeder mineralsiche Dünger bewirkt, dass nach einiger Zeit dasjenige, was auf den Feldern erzeugt wird, die mit ihm gedüngt werden, an Nährwert verliert. Das ist ein allgemeines Gesetz" (*Hence, all mineral fertilizers affect, after some time, what is grown on fields, and those (crops) lose nutritional value. This is a general law*). (Steiner, 1924: 176)

Essentially, a living soil and "living forces" were singled out as being most important for crop production, while inorganic fertilizers were identified as being responsible for quality deterioration not providing "living forces." The term "living forces" was explained by Steiner as forces surrounding organisms. Plants were described as having an "etheric aura" and animals an "etheric and astral aura" (Steiner, 1924). The "etheric aura" of plants was seen as the receiver of "living forces."

The auras and forces mentioned by Steiner are not known to science, but nevertheless there have been several attempts to visualize them. Two attempts in particular are described below, a photographic technique (Kirlian photography) to visualize the phenomenon of an aura (Kirlian, 1949) and a crystallization technique to visualize formative forces in a plant (Engqvist, 1970).

The Russian researcher Semyon Kirlian discovered in 1939 that when a plant part, e.g. a leaf, was placed on a photographic plate with a metal plate underneath and

connected to an electricity supply (high voltage, high frequency), an image was produced on the photographic plate. The image showed color fields around the electrified object that resembled an aura. Such plant images were described as revealing "the secret life of plants." However, when the method of using electricity to reveal auras around objects was subjected to further research, it was found that when high voltage enters an object electric discharge occurs, leading to ionization of water and gaseous molecules around surfaces of the object. In fact, it was found that variations in color, length, density, and curvature of the field around plants could be explained by variations in the water content on the surface and in the tissue, causing different discharge conditions (Pehek et al., 1976). Forms and colors of the electrical discharge zone around a plant are thus not an intrinsic plant quality, but a random effect of the water content, and do not represent etheric auras.

The method devised by Engqvist (1970) to illustrate "forces" in plants was based on crystallization. It involved mixing freshly pressed solution (juice) from a plant with copper chloride, placing drops of the mixture on a glass plate, and evaporating the liquid off under controlled temperature and moisture conditions. The crystallization pattern formed during evaporation was interpreted as showing the plant's organized and formative "forces." The more regular the pattern the better, with irregularity taken to indicate less organized plant "forces." From a scientific perspective, crystallization of plant solution with copper chloride during evaporation is primarily affected by the concentration of compounds in the plant juice and their reactions with copper chloride. Even if strictly controlled conditions are applied, i.e. constant temperature, moisture and rate of addition, uniform surface properties of plates, plants of similar age, etc. crystallization patterns are governed by plant solution concentrations and composition. Interpretation of crystallization patterns as being intrinsic formative "forces" of the plant, is a pseudoscientific approach.

Despite impressive scientific and technical advances, it has not been possible to identify living "forces" or "auras" surrounding plants or animals. Steiner's descriptions of such phenomena were based on his spiritual experience (Kirchmann, 1994). Note that Steiner never addressed known crop quality characteristics such as protein, carbohydrate, or fat content, or mineral or vitamin concentrations.

The statement by Steiner that our view about the function of minerals as nutrients for crops is completely wrong is not substantiated by science. Soil and plant scientists have identified 14 elements in the periodic table as being essential plant nutrients (Marschner, 2012). These nutrients are taken up as inorganic ions by roots and are involved in photosynthesis, amino acid and carbohydrate synthesis, formation of cell walls and structural components, and enzyme reactions regulating growth and reproduction of plants (Mengel et al., 2001). In stark contradiction of the statement by Steiner, supplying inorganic nutrients to crops contributes significantly to food production (Smil, 2002; Stewart et al., 2005), and has become a critical tool in

improving crop yields and quality (e.g., Cassman, 1999). Inorganic fertilizers allow plant protein content to be regulated through precision agriculture (Raun et al., 2002), increase soil fertility (e.g., Carlgren and Mattsson, 2001), and enable crops to be fortified with essential trace elements, e.g. selenium (Hartikainen, 2005).

There is no evidence to support Steiner's statement that science must "correct itself" about the nutritional role of minerals for plants. His statement (hypothesis) of "living forces" affecting crops cannot be tested, and is thus not falsifiable. However, when a hypothesis is not falsifiable, this is a sign of pseudoscience (Hines, 2003).

### Testing Balfour's and Howard's statements about mineral fertilizers

The 1940s brought the next wave of organic pioneers, with Balfour and Howard as prominent figures in the United Kingdom. In her highly influential book "*The Living Soil*," Balfour (1943) pointed out the importance of a healthy soil and the nutritional superiority of organically grown food. The central hypothesis in writings by Balfour and Howard was that there is a close relationship between soil fertility, soil humus, and human health.

According to Howard, perfectly healthy soils are the basis for health on earth: "The undernourishment of the soil is at the root of all" and health is a "birthright of life" (Howard, 1947: 12) and "Humus is the most significant of all nature's reserves" (Howard, 1947: 26). Thus, the aim of Balfour and Howard was to increase and maintain organic matter content in soils, which was regarded as a guarantee of soil health and human health. A decline in soil humus was interpreted as deterioration of soil quality and inorganic fertilizers were considered to lead to humus decreases.

Balfour stated that: "*Artificial fertilisers speed up the rate at which soil organic matter is exhausted*" (Balfour, 1943: 53). She postulated that inorganic fertilizers accelerate decomposition of organic matter in soil, leading to a decline over time. Her statement that inorganic fertilizers increase the breakdown of soil organic matter and, as a result, reduce stocks of soil organic matter has been a topic of debate since the 1940s.

Understanding of the interactions between inorganic fertilizers and soil has greatly increased since isotopic research was introduced as a tool in soil and crop science. Early nitrogen isotope studies revealed no significant impact of added mineral nitrogen fertilizer on decomposition, as microbiological activity was not stimulated and no additional carbon dioxide was released (Jansson, 1958). Studies on mineral nitrogen fertilizer and soil nitrogen interactions further corroborated that humus breakdown is not accelerated by mineral fertilizer application (Jenkinson et al., 1985).

Nevertheless, the hypothesis by Balfour that inorganic nitrogen fertilizers cause soil organic matter breakdown and reduce soil organic matter stocks is still being put forward, e.g. by Mulvaney et al. (2009). A later study concluded the interpretation by Mulvaney et al. (2009) "is false

and not supported by their data or data from numerous studies worldwide" (Powlson et al., 2010). In fact, inorganic fertilizers enhance soil humus formation. With increasing doses of nitrogen fertilizer, more roots and above-ground crop residues are produced, forming the raw material for creation of more soil organic matter (e.g., Kätker et al., 2012; Poffenbarger et al., 2017; Powlson et al., 2011). Long-term field experiments have revealed that the level of soil organic matter is the result of the production level of an agroecosystem (Johnston et al., 2009, 2017).

Recently, nutrient supply to microorganisms has been identified as influencing soil organic matter formation. It has been reported that a sufficient supply of inorganic nitrogen to microbes when decomposing organic matter consistently increases microbial growth (Spohn et al., 2016) and reduces soil organic matter decomposition (Mahal et al., 2019). Decomposition appears to be linked to the stoichiometry of nutrients available to microbes (Kirkby et al., 2013, 2014). If an optimal stoichiometric nutrient ratio for microbes is provided, for example by inorganic fertilizers, humus formation in soil can be improved by 30% (Kirkby et al., 2016). In summary, in stark contradiction to Balfour's statement, inorganic fertilizers increase organic matter in soil.

Howard (1940) considered supplying nutrients to plants through soluble inorganic fertilizers to be a "fatal error":

Artificial fertilisers were born out of the abuse of Liebig's discoveries of the chemical properties of soil. The effects of the physical properties of the soil were by-passed: its physiological life ignored, even denied, the latter a most fatal error. The essential co-partnership between the soil and the life of the creatures, which inhabit it, to which Darwin's genius had early drawn attention, is wholly forgotten. (Howard, 1947: 71–72)

Howard wrote that there is "a second method by which plants feed themselves. It is a direct connection, a kind of living bridge, between life in soil and the living portion (plants) of the soil" (Howard, 1947: 22). Howard believed that only plant nutrients made available through this second method can feed plants properly (Howard, 1947: 22–29).

Two pathways of beneficial plant-microbe interactions that resemble the living bridges referred to by Howard are symbiotic nitrogen fixation in legume roots by *Rhizobium* bacteria and symbiotic associations between plants and arbuscular mycorrhizal fungi (AMF). However, these biological symbioses can supply legumes with nitrogen and plants with phosphorus, and some micronutrients, so the entire nutrient requirement of crops is not covered. Furthermore, the contribution of AMF varies with soil conditions and systems, and the association can be both parasitic and symbiotic for crops (Ryan and Tibbett, 2008). In the present understanding, crop requirements for essential plant nutrients are met by root uptake of dissolved inorganic ions (Marschner, 2012; Mengel et al., 2001), dissolved chelated metal ions (e.g., Chen et al., 2001; Ullah and Gerzabek, 1991), and dissolved amino acids (e.g., Jones and Darrah, 1994; Näsholm et al., 2000). In summary, the statement by

Howard about living bridges by which plants are fed is incorrect.

### Testing Rusch's statements about inorganic fertilizers

Rusch aimed to apply principles that he observed nature in agricultural production, to keep farming in harmony with nature. He defined this as analogical, biological thinking, which he defined in a book about biological organic farming (Rusch, 1978). His focus was on soil microbes and their role in organic matter turnover in soil. He developed a microbial test as a tool to determine soil quality. He stated that disturbing and mixing of soil must be avoided, in order to mimic nature (Rusch, 1978: 80, 215). One major consequence of his approach is that plowing, which involves mixing soil and disturbing the formation of layers in soil, is prohibited. As a consequence, in biological organic farming organic manures may not be incorporated into the topsoil, and may only be used as a surface cover (Rusch, 1978: 158). Moreover, organic manures are considered unsuitable for the root zone, composting is considered not to be in line with nature, and decomposition of organic materials should only take place on the soil surface (Rusch, 1978: 166).

The statements by Rusch (1978) about inorganic fertilizers were as follows: "Artificial fertilization is not a normal, physiological and natural form of plant nutrition" (Rusch, 1978: 17) and "It is completely impossible to mimic the natural supply of minerals between soil and crop and this is the unavoidable error of artificial fertilization" (Rusch, 1978: 73). Rusch postulated that substances added with inorganic fertilizers are not a natural form of plant nutrition and that this type of supply is not in accordance with the physiological needs of crops. These statements question the basic understanding of plant nutrition that has prevailed since the findings by Liebig (1847) to modern times (Marschner, 2012), and require scrutiny.

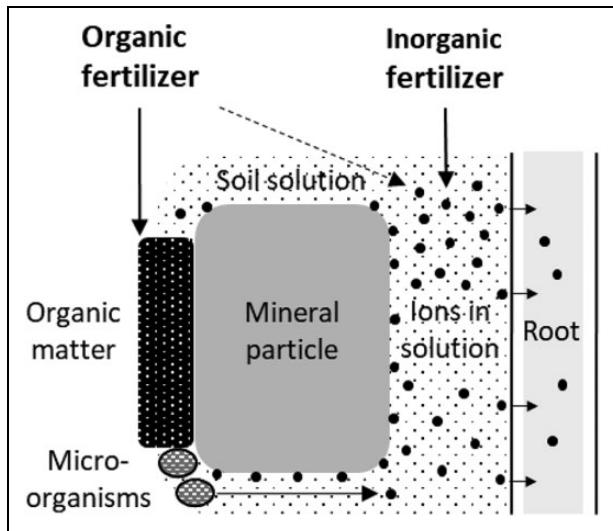
A limited number of elements in the periodic table (14) have been identified and classified as being essential plant nutrients (Mengel et al., 2001). Common compounds of these elements present in soil and water can be taken up by roots as inorganic ions in dissolved form (Marschner, 2012). Organic molecules larger than single amino acids cannot penetrate root cells. Instead, organic compounds are synthesized by plants from inorganic ions, water, and carbon dioxide. Plant roots do not discriminate between sources from which inorganic ions originate. Identical ions act physiologically alike whether added to soil with urine, slurry, manures, or compost, deposited with rainfall, supplied with irrigation water, released by soil minerals and soil organic matter, or applied with inorganic fertilizers. Furthermore, all nutrients are involved in the same processes in the soil-crop system, whether added with organic or inorganic fertilizers. For example, identical molecules such as an ammonium ion in urine undergoes the same reactions in soil as an ammonium ion added as ammonium fertilizer (Kirchmann and Pettersson, 1995).

Rusch claimed that inorganic fertilizers cannot "mimic the natural supply of minerals between soil and crop,"

where a "natural supply" means that the amount and type of plant-available nutrients are in synchrony with crop demand. This claim is not corroborated by science. Soils vary in their composition (Brady and Weil, 1996) and in their native state can be rich or deficient in nutrients and trace elements, depending on geological conditions, i.e. the bedrock from which the soils derive (Kabata-Pendia, 2000). Soils can also contain undesirable elements such as cadmium, arsenic, etc. that can affect crop composition and human health (Selinus et al., 2005). For example, there is a low content of selenium in Scandinavian soils (Eriksson et al., 2010), which can be overcome by fertilization with selenium (Hartikainen, 2005). Similarly, the zinc content is naturally very low in Turkish soils, but the problem can be overcome by adding zinc to fertilizers (Cakmak et al., 1999). Thus it is wrong to believe that soils in their natural state have a perfect nutrient composition for crop growth.

Rusch postulated that there is synchrony between nutrient release in soil and crop demand, and that the natural nutrient release in soil cannot be mimicked by inorganic fertilizer. This is an erroneous belief as such conditions rarely exist. Release of nutrients in soil is mainly controlled by temperature and moisture conditions (Kätterer and Andrén, 2001), whereas demand for nutrients is driven by crop growth (Hunt, 1982). For example, under temperate climate conditions, soils can have low temperature in spring and cannot provide sufficient nutrients when crop demand is high. In late summer, when soils are warm and moist, nutrient supply from soil can be large, but there is little or no demand by crops (Kirchmann and Bergström, 2008).

The underlying assumption by Rusch that inorganic fertilizers have an imbalanced composition and are an inferior nutrient source is also not corroborated by science. In fact, it is organic manures that are often not well adapted to crop demand. Manures produced by grazing cows exemplify this. Grazing cows deposit urine and feces on the ground and return nutrients to the site where they originated. However, nitrogen added with a urine or dung patch amounts to about  $1000 \text{ kg N ha}^{-1}$ , which is far more than surrounding plants can take up during the grazing period (Haynes and Williams, 1993; Wachendorf et al., 2005). Nitrogen in urine patches consist of around 10% organic compounds (hippuric acid, allantoin, and creatine) and 90% inorganic compounds (urea plus ammonium), causing high ammonia losses of up to 60% within a few hours (Whitehead et al., 1989). The nitrogen to phosphorus ratio, a measure of how well nutrient composition is adapted to crop need, is highly imbalanced in urine, which contains much nitrogen, but no phosphorus (Haynes and Williams, 1993). A nitrogen to phosphorus ratio of 5:1 is required to be balanced with crop demand (Sadras, 2006). Fecal material in dung patches contains mainly organically bound nitrogen (Floate and Torrance, 1970), which must undergo microbial mineralization before becoming plant available. Dung also has an imbalanced nitrogen to phosphorus ratio (1:0.25–0.4) (Haynes and Williams, 1993), not matching crop needs. While the composition of animal manures is controlled



**Figure 1.** Schematic illustration of the functions of organic and inorganic fertilizers in agricultural soils. Inorganic and organic fertilizers provide water-soluble nutrients in the form of ions dissolved in soil solution and organic fertilizers also add organic matter.

by their metabolism, inorganic fertilizers can be formulated to meet specific demands of different crops even over a growing period (through foliar or split application).

In summary, agricultural management can be inspired by nature, but mimicry of processes in natural ecosystems can be misguided. According to Rusch (1978: 90), nature is a perfect wholeness in itself. However, nature's wisdom cannot be found at the ecosystem level, but in the adaptation of individual plant and animal species (Denison, 2012).

### State-of-the-art of inorganic fertilizers

Production of fertilizers is based on the nutritional demand of higher plants (Marschner, 2012). Composition and use of inorganic fertilizers can be adapted to match different crops. The objective is to supply crops with plant nutrients in water-soluble/plant-available form instead of applying untreated minerals with low water solubility. The qualifier "artificial" or "synthetic" fertilizer gives the impression that these materials are unnatural and analogous to synthetic biocides. This is a misunderstanding as inorganic fertilizers provide essential plant nutrients comparable to organic sources. Inorganic fertilizers feed crops first of all and increase the soil nutrient pool. Organic fertilizers supply organic matter to soil (Kirchmann and Witter, 1992; Sommer et al., 2013) and to a lesser degree provide nutrients to crops (Figure 1).

Soils can be natively depleted in nutrients or have an imbalanced nutrient composition. These conditions can be overcome through the use of fertilizers, restoring and building up the depleted nutrient pool with plant-available nutrients. Using untreated minerals instead, e.g. replenishing phosphorus-depleted soil with apatite, would mean wasting a resource, as most apatite forms have no significant impact on crop production. Phosphorus content in soil would

increase, but no plant-available phosphorus would be provided and the deficiency problem would remain.

Nutrients are removed from soil through harvested products and through losses by leaching and gaseous emissions. Exclusive recycling of harvested nutrients would not compensate for losses, the balance between outtake and input would become negative and, over time, soils would become depleted lowering soil fertility. Inorganic fertilizers can compensate for losses and/or non-recycled nutrients, while organic fertilizers recycle harvested nutrients only even when redistributed.

Some key points about inorganic fertilizers:

1. Use of inorganic fertilizers means reactivation of accumulated inorganic deposits from earlier geological periods (Jansson, 1971).
2. Fertilizer production means that the chemical structure of mineral deposits is modified to become water-soluble whereby unwanted elements can be removed (Finck, 1982).
3. When nitrogen gas from the atmosphere is chemically bound to hydrogen to form ammonia in the Haber-Bosch process (Smil, 2004), the same inorganic compound is produced as in biological nitrogen fixation.
4. Inorganic N fertilizers can be produced in sustainable manner using renewable energy in the Haber-Bosch process (Ahlgren et al., 2010; Bertilsson and Kirchmann, 2021).
5. Appropriate agricultural use of inorganic N does not cause more greenhouse gas emissions than using organic manures exclusively per unit crop yield (Kirchmann et al., 2016).
6. Leaching losses of inorganic N are lower than when using organic N sources per unit crop yield (Aronsson et al., 2007; Stenberg et al., 2012; Torstensson et al., 2006).
7. The composition of inorganic fertilizers can be adapted to match the need of crops with the right type and amount of nutrients at the right time.

### Example of environmental benefit using inorganic fertilizer

Some Swedish literature references from similar organic and conventional rotations were compiled (Table 1) to illustrate how N leaching losses are affected by the exclusive use of inorganic or organic N. Supply of N in the organic rotations was exclusively through legumes (Aronsson et al., 2007; Stenberg et al., 2012; Torstensson et al., 2006). Losses of N leaching were expressed per hectare and unit yield. Yield differences between systems were similar to those reported in Swedish statistics (Kirchmann, 2019). In two studies, leaching of N per hectare was somewhat higher from conventional rotations (38 vs 34 and 13 vs 11 kg N ha<sup>-1</sup> yr<sup>-1</sup>) but lower in the third study (5 vs 19 kg N ha<sup>-1</sup> yr<sup>-1</sup>). However, the amount of N leached per product revealed the true environmental impact. Expressing leaching losses per unit product (Table 2), showed that organic rotations resulted in

**Table 2.** Leaching losses of N from comparative crop rotations without livestock using inorganic N (conventional) and exclusively legume N (organic).

Reference and system	Mean yield over the rotation (kg dm ha <sup>-1</sup> yr <sup>-1</sup> )	Mean N leaching over the rotation (kg N ha <sup>-1</sup> yr <sup>-1</sup> )	Mean N leaching per yield (kg N Mg <sup>-1</sup> yield)
<b>Torstensson et al., (2006)</b>			
6-year conventional rotation (N as mineral fertilizer)	5783	38	6.6
6-year organic rotation (N as legumes in rotation)	2926	34	11.6
<b>Aronsson et al., (2007)</b>			
6-year conventional rotation (N as mineral fertilizer)	5580	13	2.3
6-year organic rotation (N as legumes in rotation)	3090	11	3.6
<b>Stenberg et al., (2012)</b>			
5-year conventional rotation (N as mineral fertilizer)	6100	5	0.8
5-year organic rotation (N as legumes in rotation)	2440	19	7.8

consistently higher leaching losses ranging from 3.6 to 11.6 kg N Mg<sup>-1</sup> organic yield as compared to 0.8 to 6.6 kg N Mg<sup>-1</sup> conventional yield. In fact, leaching losses per unit product were 78–69% lower in the conventional rotations. These measures indicate that environmental leaching loads of N are increased by organic farming when the boundary condition to produce similar amount of crops is applied.

## Concluding remarks

Organic farming was founded on the doctrine the inorganic fertilizers should not be used, in the belief that they are harmful for soil, food quality, and human health. Statements about inorganic fertilizers made by the initiators of organic farming reflect their understanding of nature. The reasons they gave for banning inorganic fertilizers could not be verified at the time, due to poor understanding of nutrient turnover in soil-plant systems, insufficient experience, and lack of scientific methodology. Understanding of soil-plant systems has greatly improved since the 1950s, however, so stringent analysis of the original arguments is now possible.

This review and analysis of core statements by the initiators of the organic farming movement showed that their stated reasons for prohibiting inorganic fertilizers are not corroborated by scientific facts. Their decision criteria were found to have no scientific legitimacy, and thus exclusion of inorganic fertilizers cannot be categorized as a principle based on science. Modern science requires invalid hypotheses to be rejected and false concepts to be corrected, so if theories and principles are maintained despite lack of evidence, science is ignored. If non-valid theories/concepts are not rejected upon falsification, they become articles of faith and make matters of science into matters of doctrine. Once a concept has become a doctrine, revision is often regarded as irrelevant.


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## ORCID iD

Holger Kirchmann  <https://orcid.org/0000-0003-4779-412X>

## References

- Ahlgren S, Bernesson S, Nordberg Å, et al. (2010) Nitrogen fertilizer production based on biogas—energy input, environmental impact and land use. *Bioresource Technology* 101: 7181–7184.
- Aronsson H, Torstensson G and Bergström L (2007) Leaching and crop uptake of N, P, and K in a clay soil with organic and conventional cropping systems on a clay soil. *Soil Use and Management* 23: 71–81.
- Balfour EA (1943) *The Living Soil: Evidence of the Importance to Human Health of Soil Vitality, with Special Reference to Post-war Planning*. London: Faber & Faber Ltd. London, U.K., 276 pp.
- Bertilsson GOB and Kirchmann H (2021) Sustainable N fertilizer production through the combination: crop residues – biogas – ‘Haber-Bosch’. Analysis of a straw-nitrogen loop. *Agricultural Systems* 190: 103100.
- Brady N and Weil R (1996) *The Nature and Properties of Soils*, 11th edn. Hoboken, NJ: Prentice Hall, 740 pp.
- Cakmak I, Kalayci M, Ekiz H, et al. (1999) Zinc deficiency as a practical problem in plant and human nutrition in Turkey: a NATO-science for stability project. *Field Crops Research* 60: 175–188.
- Carlgrén K and Mattsson L (2001) Swedish soil fertility experiments. *Acta Agriculturae Scandinavica, Section B, Soil and Plant Science* 5: 49–78.
- Cassman KG (1999) Ecological intensification of cereal production systems: yield potential, soil quality, and precision agriculture. *Proceedings of the National Academy of Science U.S.A.* 96: 5952–5959.
- Chen Y, Magen H and Clapp CE (2001) *Plant Growth Stimulation by Humic Substances and Their Complexes With*



- Iron. York: International Fertilizer Society 14 pp. [Proceedings No 470].
- Denison RF (2012) *Darwinian Agriculture*. Princeton, NJ: Princeton University Press, 249 pp.
- Engqvist M (1970) *Gestaltkräfte des Lebendigen*. Germany: Vittorio Klostermann Verlag.
- Eriksson J, Mattsson L and Söderström M (2010) *Current status of Swedish Arable Soils and Crops. Data From the Period 2001-2007 [in Swedish]*, Report No. 6349. Stockholm: Swedish Environmental Protection Agency.
- Finck A (1982) *Fertilizers and Fertilization: Introduction and Practical Guide to Crop Fertilization*. Germany: Verlag Chemie, 438 pp.
- Floate MJS and Torrance CJW (1970) Decomposition of the organic materials from hills and pastures. 1. Incubation method for studying the mineralisation of carbon, nitrogen and phosphorus. *Journal of the Science of Food and Agriculture* 21: 116–120.
- Granstedt A (2000) Increasing the efficiency of plant nutrient recycling within the agricultural system as a way of reducing the load to the environment—experience from Sweden and Finland. *Agriculture, Ecosystems & Environment* 80: 169–185.
- Granstedt A, Schneider T, Seuri P, et al. (2008) Ecological recycling agriculture to reduce nutrient pollution to the Baltic Sea. *Biological Agriculture & Horticulture* 26: 279–307.
- Hartikainen H (2005) Biochemistry of selenium and its impact on food chain quality and human health. *Journal of Trace Elements and Human Health* 18: 309–318.
- Haynes RJ and Williams PH (1993) Nutrient cycling and soil fertility in the grazed pasture ecosystem. *Advances in Agronomy* 49: 119–199.
- Hines T (2003) *Paranormal and the Pseudoscience*, 2nd edn. New York, NY: Prometheus Books, 485 pp.
- Howard A (1940) *My Agricultural Testament*. Oxford: Oxford University Press, 253 pp.
- Howard A (1947) *The Soil and Health. A Study of Organic Agriculture*. New York, NY: The Devin-Adair Company, 307 pp.
- Hunt R (1982) *Plant Growth Curves – The Functional Approach to Plant Growth Analysis*. Great Britain: E. Arnold Limited, 248 pp.
- Jansson SL (1958) Tracer studies on nitrogen transformation in soil with special attention to mineralization-immobilization relationship. *Annals of Royal Agricultural College of Sweden* 24: 101–361.
- Jansson SL (1971) The naturalness of commercial fertilizers. An ecological treatise. *Acta Agralia Fennica* 123: 173–185.
- Jenkinson DS, Fox RH and Rayner JH (1985) Interactions between fertilizer nitrogen and soil nitrogen—the so called ‘priming’ effect. *Journal of Soil Science* 36: 425–444.
- Johnston AE, Poulton PR and Coleman K (2009) Soil organic matter: its importance in sustainable agriculture and carbon dioxide fluxes. *Advances in Agronomy* 101: 1–57.
- Johnston AE, Poulton PR, Coleman K, et al. (2017) Changes in soil organic matter over 70 years in continuous arable and ley-arable rotations on a sandy loam soil in England. *European Journal of Soil Science* 68: 305–316.
- Jones DL and Darrah PR (1994) Amino acid influx at the soil-root interface of *Zea mays* L. and its implications in the rhizosphere. *Plant Soil* 163: 1–12.
- Kabata-Pendia A (2000) *Trace Elements in Soils and Plants*. 3rd edn. Boca Raton, FL: CRC Press, 413 pp.
- Kätterer T and Andrén O (2001) The ICBM family of analytically solved models of soil carbon, nitrogen and microbial biomass dynamics—descriptions and application examples. *Ecological Modelling* 136: 191–207.
- Kätterer T, Bolinder MA, Berglund K, et al. (2012) Strategies for carbon sequestration in agricultural soils in northern Europe. *Acta Agriculturae Scandinavica, Animal Science Section A* 62: 181–198.
- Kirchmann H (1994) Biologic dynamic farming—an occult form of alternative agriculture? *Journal of Agricultural and Environmental Ethics* 7: 173–187.
- Kirchmann H (2019) Why organic farming is not the way forward. *Outlook on Agriculture* 48: 22–27.
- Kirchmann H and Bergström L (2008) *Organic Crop Production—Ambitions and Limitations*. Dordrecht: Springer, 240 pp.
- Kirchmann H and Pettersson S (1995) Human urine—chemical composition and fertilizer use efficiency. *Fertilizer Research* 40: 149–154.
- Kirchmann H and Witter E (1992) Composition of fresh, aerobic and anaerobic farm animal dung. *Bioresource Technology* 40: 137–142.
- Kirchmann H, Kätterer T, Bergström L, et al. (2016) Flaws and criteria for design and evaluation of comparative organic and conventional cropping systems. *Field Crops Research* 186: 99–106.
- Kirkby CA, Richardson AE, Wade LJ, et al. (2013) Carbon-nutrient stoichiometry to increase soil-carbon sequestration. *Soil Biology & Biochemistry* 60: 77–86.
- Kirkby CA, Richardson AE, Wade LJ, et al. (2014) Nutrient availability limits carbon sequestration in arable soils. *Soil Biology & Biochemistry* 68: 402–409.
- Kirkby CA, Richardson AE, Wade LJ, et al. (2016) Inorganic nutrients increase humification efficiency and C sequestration in an annually cropped soil. *PLoS ONE* 11(5): e0153698.
- Kirlian SD (1949) Method for receiving photographic pictures of different types of objects. Patent N106401, USSR.
- Koepf HH (1973) Organic management reduces leaching of nitrate. *Biodynamics* 108: 20–30.
- Liebig J (1847) *Chemistry in its Application to Agriculture and Physiology. From the Last London Edition, Much Improved*. Philadelphia, PA: T.B. Peterson, USA.
- Mahal NK, Osterholz WR, Miguez FE, et al. (2019) Nitrogen fertilizer suppresses mineralization of soil organic matter in maize agroecosystems. *Frontiers in Ecology and Evolution* 7: 59.
- Marschner P (2012) *Mineral Nutrition of Higher Plants*, 3rd edn. London: Academic Press, Elsevier Ltd., 651 pp.
- Mengel K, Kirkby EA, Kosegarten H, et al. (2001) *Principles of Plant Nutrition*, 5th edn. Dordrecht, NL: Springer Science+Business Media, 849 pp.
- Muller A, Schader C, El-Hage Scialabba N, et al. (2017) Strategies for feeding the world more sustainably with organic agriculture. *Nature Communications* 8: 1290.
- Mulvaney RL, Khan SA and Ellsworth TR (2009) Synthetic nitrogen fertilizers deplete soil nitrogen: a global dilemma for sustainable cereal production. *Journal of Environmental Quality* 38: 2295–2314.



- Näsholm T, Huss-Danell K and Högberg P (2000) Uptake of organic nitrogen in the field by four agriculturally important plant species. *Ecology* 81: 1155–1161.
- Pehek JO, Kyler HJ and Faust DL (1976) Image modulated corona discharge photography. *Science* 194: 263–270.
- Poffenbarger HJ, Barker DW, Helmers MJ, et al. (2017) Maximum soil organic carbon storage in Midwest US cropping systems when crops are optimally nitrogen-fertilized. *PLoS ONE* 12: e0172293.
- Powlson DS, Jenkinson DS, Johnston AE, et al. (2010) Comments on “Synthetic nitrogen fertilizers deplete soil nitrogen: a global dilemma for sustainable cereal production,” by R.L. Mulvaney, S.A. Khan, and T.R. Ellsworth in the *Journal of Environmental Quality* 2009 38:2295–2314. *Journal of Environmental Quality* 39: 749–752.
- Powlson DS, Whitmore AP and Goulding KWT (2011) Soil carbon sequestration to mitigate climate change: a critical re-examination to identify the true and the false. *European Journal of Soil Science* 62: 42–55.
- Popper K (1959) *The Logic of Scientific Discovery*. London: Hutchinson & Co, 544 pp.
- Raun WR, Solie JB, Johnson GV, et al. (2002) Improving nitrogen use efficiency in cereal grain production with optical sensing and variable rate application. *Agronomy Journal* 94: 815–820.
- Rusch HP (1978) *Bodenfruchtbarkeit. Eine Studie biologischen Denkens (Soil Fertility. A Case of Biological Reasoning)* [in German], 3rd Printing. Heidelberg, Germany: Haug Verlag, 243 pp.
- Ryan MH and Tibbett M (2008) The role of arbuscular mycorrhizas in organic farming. In: Kirchmann H. and Bergström L. (eds) *Organic Crop Production—Ambitions and Limitations*. Germany: Springer Science, pp. 189–229.
- Sadras VO (2006) The N:P stoichiometry of cereal, grain legume and oilseed crops. *Field Crops Research* 95: 13–29.
- Selinus O, Alloway B, Centeno JA, et al. (2005) *Essentials of Medical Geology—Impacts of Natural Environment on Public Health*. Amsterdam: Elsevier Academic Press, 812 pp.
- Smil V (2002) Nitrogen and food production: Proteins for human diets. *Ambio* 31: 126–131.
- Smil V (2004) *Enriching the Earth. Fritz Haber, Carl Bosch and The Transformation of World Food Production*. Cambridge, MA: MIT Press, 353 pp.
- Smith LG, Kirk GJD, Jones PJ, et al. (2019) The greenhouse gas impacts of converting food production in England and Wales to organic methods. *Nature Communications* 10: 4641.
- Sommer SG, Christensen ML, Schmidt T, et al. (2013) *Animal Manure Recycling*. Hoboken, NY: Wiley Ltd, 364 pp.
- Spohn M, Pötsch EM, Eichorst SA, et al. (2016) Soil microbial carbon use efficiency and biomass turnover in a long-term fertilization experiment in a temperate grassland. *Soil Biology & Biochemistry* 97: 168–175.
- Steiner R (1924) *Geisteswissenschaftliche Grundlagen zum Gedeihen der Landwirtschaft (Spiritual Foundations for the Renewal of Agriculture)* (in German). 5th Printing 1975, Rudolf Steiner Nachlassverwaltung, Dornach, Switzerland, 256 pp.
- Stenberg M, Ulén B, Söderström M, et al. (2012) Tile drain losses of nitrogen and phosphorus from fields under integrated and organic crop rotations. A four-year study on a clay soil in southwest Sweden. *Science of the Total Environment* 434: 79–89.
- Stewart WM, Dibb DW, Johnston AE, et al. (2005) The contribution of commercial fertilizer nutrients to food production. *Agronomy Journal* 97: 1–6.
- Torstensson G, Aronsson H and Bergström L (2006) Nutrient use efficiencies and leaching of organic and conventional cropping systems in Sweden. *Agronomy Journal* 98: 603–615.
- Ullah SM and Gerzabek MH (1991) Influence of fulvic and humic acids on Cu-toxicity and V-toxicity to *Zea-Mays* (L.). *Bodenkultur* 42: 123–134.
- Wachendorf C, Traube F and Wachendorf M (2005) Nitrogen leaching from 15 N labelled cow urine and dung applied to grassland on a sandy soil. *Nutrient Cycling in Agroecosystems* 73: 89–100.
- Whitehead DDR, Lockyer DR and Raistrick N (1989) Volatilization of ammonia from urea applied to soil: influence of hippuric acid and other constituents of livestock urine. *Soil Biology & Biochemistry* 21: 803–808.